
MIXTURE PROBLEMS – RADIATOR

□ INTRODUCTION

Suppose a 12-liter radiator is filled with an antifreeze/water solution at a 25% concentration of antifreeze. In that case, 25% of the 12 liters is antifreeze, while the rest (75%) is water. Multiplying 12 liters by 25% shows that 3 liters of the solution are antifreeze, and the other 9 liters (12 liters $\times$ 75%) are water. Thus, the key formula we need for the percent mixture problems in this chapter is



$\text{Quantity} \times \text{Concentration} = \text{Actual Amount}$
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□ EXAMPLE

A 10-liter radiator is filled with 42% antifreeze solution. How much of the antifreeze solution must be drained and replaced with pure antifreeze to bring the concentration of the final solution up to 58%?

Solution: We'll set up all the info in a chart, utilizing the above formula:

	Quantity (in liters)	$\times$ Concentration (of antifreeze)	= Actual Amount (of antifreeze)
Original	10	42%	4.2
Drain	x	42%	$0.42x$
Replace	x	100%	x
Final	10	58%	5.8

Notice that the four rows in our chart represent each of the problem's four phases.

In the statement of the problem, we are given that it's a 10-liter radiator filled with 42% antifreeze. The product of these two numbers is 4.2, as per the formula given above.

The question asks us how much to remove -- we don't know, so this is where the variable x comes in. What is the concentration of the x liters we're to remove? It's the same as the concentration of the entire 10 liters: 42%. And, of course, the product of x with 42% is $0.42x$.

Now for the 3rd row of the chart. The number of liters of pure antifreeze needed to refill the radiator must equal the number of liters drained off, namely x . And the refill fluid is pure antifreeze, which has a concentration of 100%; hence, the last entry in the row is $x \times 100\%$, or simply x .

The bottom row: The radiator has been refilled, so it contains 10 liters of solution, just as in the original row. The concentration of this final mixture is stipulated to be 58%, and the product is clearly 5.8.

Thus, looking at the right-hand column of the chart (Actual Amount), we can say that we started with 4.2 liters of antifreeze, subtracted $0.42x$ liters, added x liters, and ended up with 5.8 liters of antifreeze. Now we translate these thoughts into an equation:

$$4.2 - 0.42x + x = 5.8$$

$$\Rightarrow 0.58x = 1.6$$

$$\Rightarrow x = 2.76, \text{ and thus}$$

2.76 liters should be drained out

Note: If the antifreeze solution needs to be diluted, then we drain a certain amount of the solution and replace it with pure water, which has an antifreeze concentration of 0%.

Homework

1. A 9-liter radiator is filled with 67% antifreeze solution. How much of the antifreeze solution must be drained and replaced with pure antifreeze to bring the concentration of the final mixture up to 95% antifreeze?
2. A 5-liter radiator is filled with 58% antifreeze solution. How much of the antifreeze solution must be drained and replaced with pure water to bring the concentration of the final mixture down to 25% antifreeze?
3. A 4-liter radiator is filled with 32% antifreeze solution. How much of the antifreeze solution must be drained and replaced with pure antifreeze to bring the concentration of the final mixture up to 75% antifreeze?
4. A 20-liter radiator is filled with 75% antifreeze solution. How much of the antifreeze solution must be drained and replaced with pure water to bring the concentration of the final mixture down to 23% antifreeze?

Solutions

1. 7.64 liters
2. Note: The concentration of antifreeze in pure water is 0%.
 $5(.58) - .58x + 0x = 5(.25) \Rightarrow 2.84 \text{ liters}$
3. 2.53 liters 4. 13.87 liters